

# Polymeric Nanoencapsulation of Indocyanine Green for Photodynamic Therapy Technique

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## Abstract

Indocyanine green (ICG) has been exploited as a photosensitizer for use in cancerous phototherapy including breast, brain, and skin tumors. Although ICG is of particular advantage for use in cancer phototherapy, it adversely tends to disintegrate in aqueous medium and such degradation can be markedly accelerated by light irradiation (photodegradation) and/or heating (thermal degradation). Furthermore, ICG after administered intravenously will be readily bound with blood proteins and hence leads to only  $2\pm 4$  min of plasmatic half-life. Among various pharmaceutical polymers, poly (lactic-co-glycolic acid) (PLGA), PEG, Chitosan are the best defined biomaterials with FDA approval for drug encapsulation due to its biocompatibility, biodegradability, and controllability for drug release. This review shows that novel applications of biodegradable polymers as nanocarrier for ICG to improve of physicochemical properties of ICG such as Half-life.

**Keywords:** Biodegradable polymers, Indocyanine green (ICG), Photodynamic therapy (PDT), Cancer treatment.

## Introduction

As mentioned earlier, the encapsulation of ICG in the polymeric construct enhances its physicochemical attributes such as half-life, bioavailability, stability and pharmacokinetic characteristics, and reduces the ease of degradation. Moreover, these versatile ICG-encapsulated NIR theranostic nanoparticles also offer numerous other advantages such as significant targeting ability and sensitivity in tumor imaging, which facilitate them in exploring other innovative therapeutic applications. Herein, we elaborate on the discussion of ICG encapsulation and its delivery using various ground-breaking polymer nanocomposites with a set of examples. Among various approaches of breast cancer treatment, near-infrared (NIR)-mediated phototherapy is one of the most promising strategies for serving as a supplement to traditional cancer therapies since it can provide 1) enhanced tissue penetration efficacy as compared with that operated by visible light and 2) moderate toxicity to normal cells/tissues through use of targeted photosensitive agents and/or spatially controlled light irradiation [1]. Generally speaking, phototherapy is carried out by hyperthermia and/or reactive oxygen species (ROS) generated from the photosensitizers under light illumination in the presence of oxygen that the former may cause thermal ablation of cancer cells (i.e., photothermal therapy; PTT), while the latter may seriously interfere cellular metabolism and thus trigger programmed cell death (i.e., photodynamic therapy; PDT) [2,3]. No matter which mechanism is utilized, the photosensitizer plays a key role in the effectiveness of phototherapy. Indocyanine green (ICG) is an U.S. Food and Drug Administration (FDA)-approved tricarbocyanine dye which

## Article Information

**Article Type:** Review

**Article Number:** JBRR123

**Received Date:** 01 May, 2019

**Accepted Date:** 10 May, 2019

**Published Date:** 17 May, 2019

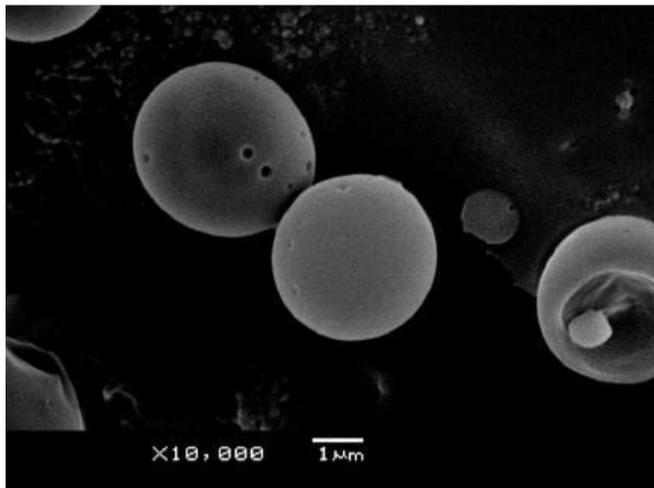
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**Citation:** Saboktakin M (2019) Polymeric Nanoencapsulation of Indocyanine Green for Photodynamic Therapy Technique. J Biomed Res Rev Vol: 2, Issu: 1 (62-76).

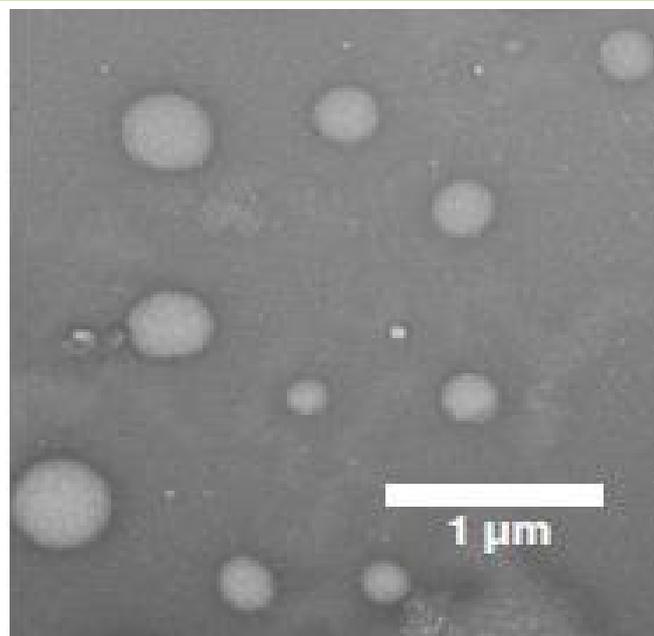
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enables to absorb and fluoresce in the region of 650±850 nm. Currently, in addition to serving as a fluorophoric agent for use in diagnostic purposes such as NIR image-guided oncologic surgery [4], fluorescence angiography [5], and lymph node detection of cancer [6], ICG has been exploited as a photosensitizer for use in cancerous phototherapy including breast, brain, and skin tumors [7-9] since it enables to produce heat and ROS (i.e., singlet oxygen) upon NIR irradiation. Although ICG is of particular advantage for use in cancer phototherapy, it adversely tends to disintegrate in aqueous medium and such degradation can be markedly accelerated by light irradiation (photodegradation) and/or heating (thermal degradation) [10]. Furthermore, ICG after administered intravenously will be readily bound with blood proteins and hence leads to only 2±4 min of plasmatic half-life [11,12]. These circumstances seriously hinder the applicability of ICG in the clinic and thus a strategy that enables to enhance the aqueous stability and target efficiency of ICG is certainly needed for ICG-mediated therapy. Nanomedicine may offer a feasible means for usage of ICG without aforementioned defects since it may provide merits of enhanced bioavailability, improved stability, and security for the payload [13]. In terms of the materials used for making drug carrier, polymer is often considered as the preferred candidate since it can be manipulated to tailor the properties and/or functionalities required by the product [14]. Among various pharmaceutical polymers, poly (lactic-co-glycolic acid) (PLGA) is the copolymer of poly(lactic acid) and poly(glycolic acid) and is one of the best defined biomaterials with FDA approval for drug encapsulation due to its biocompatibility, biodegradability, and controllability for drug release [15]. Polyethylene glycol (PEG), another FDA-approved polymer with characteristics of nontoxicity and less immunogenicity, is frequently used for surface modification of drug carrier since the retention time of the PEG-coated particle in the blood circulation can be markedly increased [16]. Taken all together, we aim to develop an anti-HER2 ICG-encapsulated PEG-coated PLGA nanoparticles (HIPPNPs) for targeted phototherapy of HER2-expressing breast cancer cells. The use of ICG by implantation of HIPPNPs instead of naked molecules is advantageous because the polymeric carrier (i.e., HIPPNP) may 1) potentially protect the entrapped ICG from degradation caused by external stimuli such as light, heat, and/or extreme pH [17]; 2) precisely localize the therapeutic region to reduce off-target toxicity, and 3) provide accurate estimation for the efficacy of ICG-mediated phototherapy. There are many reports that ICG encapsulators, such as (poly (lactic-co-glycolic acid) (PLGA) nanoparticles (diameter~360 nm) and silica-polymer composite microcapsules (diameter ~0.6 to 2 µm) improve the molecular instability of ICG and prolong its plasma half-life [18,19]. However, both of these nanoparticles are limited in size for *in vivo* tumor imaging depending on their EPR effects. Recently, several publications have reported promising results using smaller nanoparticles to encapsulate ICG for *in vivo* imaging. For example, Zheng et al. [20] developed ICG encapsulated PLGA-lipid nanoparticles conjugated with folic acid (FA) and demonstrated their use as NIR contrast agents for tumor diagnosis and targeted imaging. Altinoglu et al. also synthesized biodegradable

calcium phosphosilicate nanoparticles (CPNPs) and demonstrated that small size (16 nm) ICG-encapsulating CPNPs have significantly better contrast agent optical properties than free fluorophores for tumor imaging [21]. Other inorganic delivery systems using silica nanoparticles have been developed to encapsulate ICG, and the ICG-SiO<sub>2</sub> nanoparticles have the potential to be used as contrast agents for optical NIR imaging as well [22]. Among these nanocarriers, micelles are one of the successful types of drug delivery systems for *in vivo* applications due to their small size (approximately 10-100 nm), which reduce clearance by the reticuloendothelial system (RES) and allow for an enhanced EPR effect [23,24]. Therefore, the encapsulation and stabilization of ICG dye as a contrast agent in micellar systems is of particular interest. For example, Pluronic F-127 (PF-127) polymeric micelles are approved by the FDA and have been successfully demonstrated to encapsulate and stabilize ICG as an NIR contrast agent for optical imaging [25,26]. Encapsulation of ICG within various micellar systems was also investigated by Kirchherr and co-workers, and they found many micellar systems improved the optical properties and stability of the ICG [27]. More interestingly, Zheng et al. have recently reported a dual-functional ICG-PL-PEG agent with several unique features for optical imaging and photo-therapy [28]. This may emerge as a new strategy for combining tumor treatment and diagnosis together, using nanovectors with ICG. In summary, this editorial discussed recent developments in nanocarrier ICG contrast agents for NIR optical imaging. Here just some of the areas are collected in terms of subjects and interests but it is hoped that every reader will find something of interest to them. Manchanda et al. [29] have recently reported the fabrication of a novel polymer nanoparticle delivery system with simultaneously entrapped indocyanine green (ICG) and doxorubicin (DOX). This system has potential applications for combined chemotherapy and hyperthermia. Research in our group showed that simultaneous use of ICG and DOX with localized hyperthermia can produce the same effect as that achieved by larger doses of chemotherapy alone. In this study, we explored the potential of dual-agent PLGA nanoparticles (ICG-DOXPLGANPs) to overcome multidrug resistance (MDR) mechanisms in cancer cells by increasing intracellular drug concentrations via nanoparticle uptake. ICG-DOX- PLGANPs were prepared by the O/W emulsion solvent evaporation method. The dominant processing parameters that control particle size and drug entrapment efficiencies of ICG and DOX were PLGA concentration, PVA concentration and initial drug content. We optimized our previous formulation based on those parameters. Entrapment efficiency of the optimized ICG- DOX-PLGANPs was measured by fluorescence measurements using the DMSO burst release procedure. The internalization of ICG-DOX- PLGANPs by three cancer cell lines was visualized by confocal laser microscopy and fluorescence microscopy. Cytotoxicity was assessed using the SRB assay. The nanoparticles produced by optimal formulation had sizes of 135±2 nm, (n=3) with a low poly-dispersity index (0.149±0.014, n=3) and a zeta potential of -11.67±1.8 mV. Drug loading was approximately 3% w/w for ICG and 4% w/w for DOX (n=3). Cellular uptake of ICG and DOX from



**Figure 1:** Observation of PLGA microspheres by SEM.



**Figure 2:** TEM image of presenting of morphology of PIN.

ICG-DOX-PLGANPs in DOX-resistant MESSA/Dx5 cancer cells was higher compared to free ICG and free DOX treatment. However, the same phenomenon was not observed in MES-SA and SKOV-3 cancer cell lines. The SRB cytotoxicity results show that ICG-DOX-PLGANPs are more toxic than free DOX in DOX-resistant cell lines. In the development of drugs for intra-articular administration, sustained-release formulations are desirable because it is difficult to maintain the effect of conventional injections due to immediate drug leakage from the joint cavity. In this study, a sustained-release poly(lactic-co-glycolic acid) (PLGA) microsphere formulation for intra-articular administration containing indocyanine green (ICG) as a model drug was prepared to follow its fate after intra-articular administration in rats with a real-time *in-vivo* imaging system. ICG administered as an aqueous solution leaked from the joint cavity in a short time and was excreted outside the body within 1–3 d. However, ICG in the sustained-release formulation was retained in the joint cavity and released for 2 weeks. Next, a sustained-release formulation

containing PLGA microspheres in a hyaluronic acid (HA) gel formulation was prepared [30]. Figure 1

Ronak et al. [31] have been developed and characterized multifunctional biodegradable and biocompatible poly lactic-co-glycolic acid (PLGA) nanoparticles loaded with indocyanine green (ICG) as an optical-imaging contrast agent for cancer imaging and as a photothermal therapy agent for cancer treatment. PLGA-ICG nanoparticles (PIN) were synthesized with a particle diameter of  $246 \pm 11$  nm, a polydispersity index of  $0.10 \pm 0.03$ , and ICG loading efficiency of

$48.75 \pm 5.48\%$ . PIN were optically characterized with peak excitation and emission at 765 and  $810 \pm 5$  nm, a fluorescence lifetime of  $0.30 \pm 0.01$  ns, and peak absorbance at 780 nm. The cytocompatibility study of PIN showed 85% cell viability till 1-mg/ml concentration of PIN.

Successful cellular uptake of ligand conjugated PIN by prostate cancer cells (PC3) was also obtained. Both phantom-based and *in vitro* cell culture results demonstrated that PIN (1) have the great potential to induce local hyperthermia (i.e., temperature increase of 8 to  $10^\circ\text{C}$ ) in tissue within 5 mm both in radius and in depth; (2) result in improved optical stability, excellent biocompatibility with healthy cells, and a great targeting capability; (3) have the ability to serve as an image contrast agent for deep-tissue imaging in diffuse optical tomography. Figure 2

Nowadays, a new technique such as photodynamic therapy (PDT) is used to achieve effective root canal disinfection and eliminate *Enterococcus faecalis* as the most prevalent species associated with secondary endodontic infections and treatment failures. Employment of an optimized nontoxic photosensitizer (PS) such as indocyanine green (ICG) is a crucial part of this technique; the current study aimed at improving ICG photodynamic properties through conjugation of ICG into nano-graphene oxide (nGO) as a new PS, to evaluate the antimicrobial effects of nGO/ICG against *E. faecalis*. The PDT based on ICG, at the concentration loaded in nGO, showed a significant reduction in the number of *E. faecalis* at a lower concentration of ICG. In conclusion, the application of nGO as a new drug delivery system, in addition to the anti-bacterial property, offers other benefits such as cost beneficial outcomes due to using the lower dye concentration (less toxicity), and less tooth discoloration. The nGO/ICG-PDT could be proposed as a new approach to treat endodontic infections, alone or in combination with conventional root canal treatments [32]. Figure 3

Zheng et al. [33] had successfully constructed the fluorescence image guided photothermal therapy reagents based on (WO+ICG)@PLGA nanoparticles. In our design, to improve their tumor targeting, the macrophages as cell-based biocarriers were employed for delivery the (WO+ICG)@PLGA nanoparticles. The macrophages carried these nanoparticles still had phagocytosis to tumor cells and could also secrete plenty of anti-tumor cytokines for immunotherapy of carcinoma. They also further elucidated the superior solid tumor suppression efficiency of the (WO+ICG)@PLGA nanoparticles loaded macrophages targeting biocarriers delivery system *in*



Figure 3: The SEM image of the nGO.

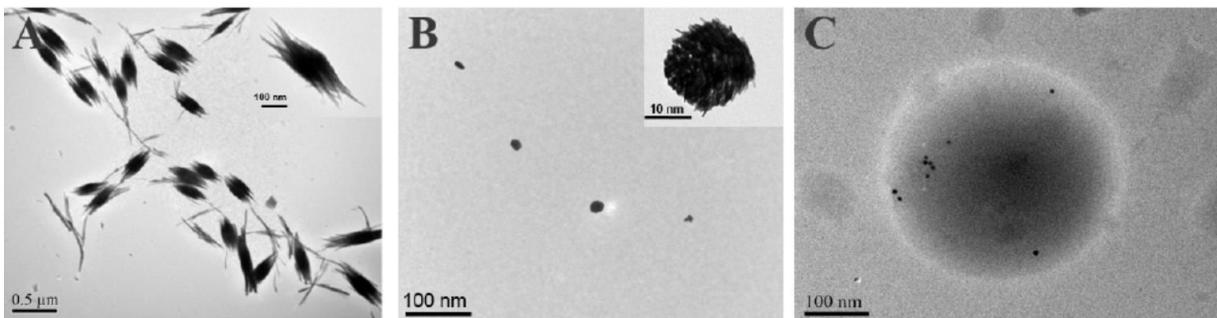


Figure 4: The TEM images of (A) Nanorod-Bundled WO, (B) Granulated.

*in vivo*. The system achieved a significant antitumor effect by activating immunotherapy and photothermal therapy *in vivo*. Hence, such kind of (WO+ICG)@PLGA nanoparticles loaded macrophages delivery system has great potential applications as targeting biocarriers loading drugs and imaging agents for visual-guided synergistic therapy *in vivo*. Figure 4

The photothermal heating and release properties of biocompatible organic nanoparticles, doped with a near infrared croconaine (Croc) dye, were compared with analogous nanoparticles doped with the common near-infrared dyes ICG and IR780. Separate formulations of lipid-polymer hybrid nanoparticles and liposomes, each containing Croc dye, absorbed strongly at 808 nm and generated clean laser-induced heating (no production of  $^1O_2$  and no photobleaching of the dye). In contrast, laser-induced heating of nanoparticles containing ICG or IR780 produced reactive  $^1O_2$ , leading to bleaching of the dye and also decomposition of coencapsulated payload such as the drug doxorubicin. Croc dye was especially useful as a photothermal agent for laser-controlled release of chemically sensitive payload from nanoparticles. Solution state experiments demonstrated repetitive fractional release of water-soluble fluorescent dye from the interior of thermosensitive liposomes. Additional experiments used a focused laser beam to control leakage from immobilized

liposomes with very high spatial and temporal precision. The results indicate that fractional photothermal leakage from nanoparticles doped with Croc dye is a promising method for a range of controlled release applications [34]. Figure 5

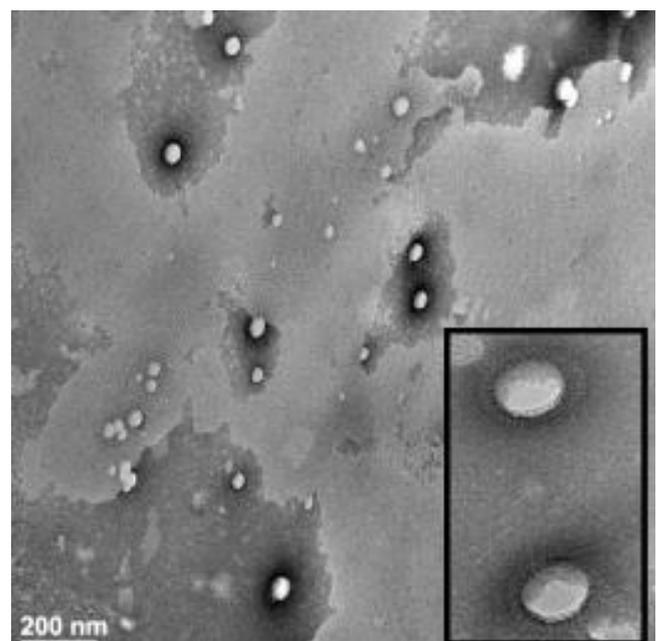
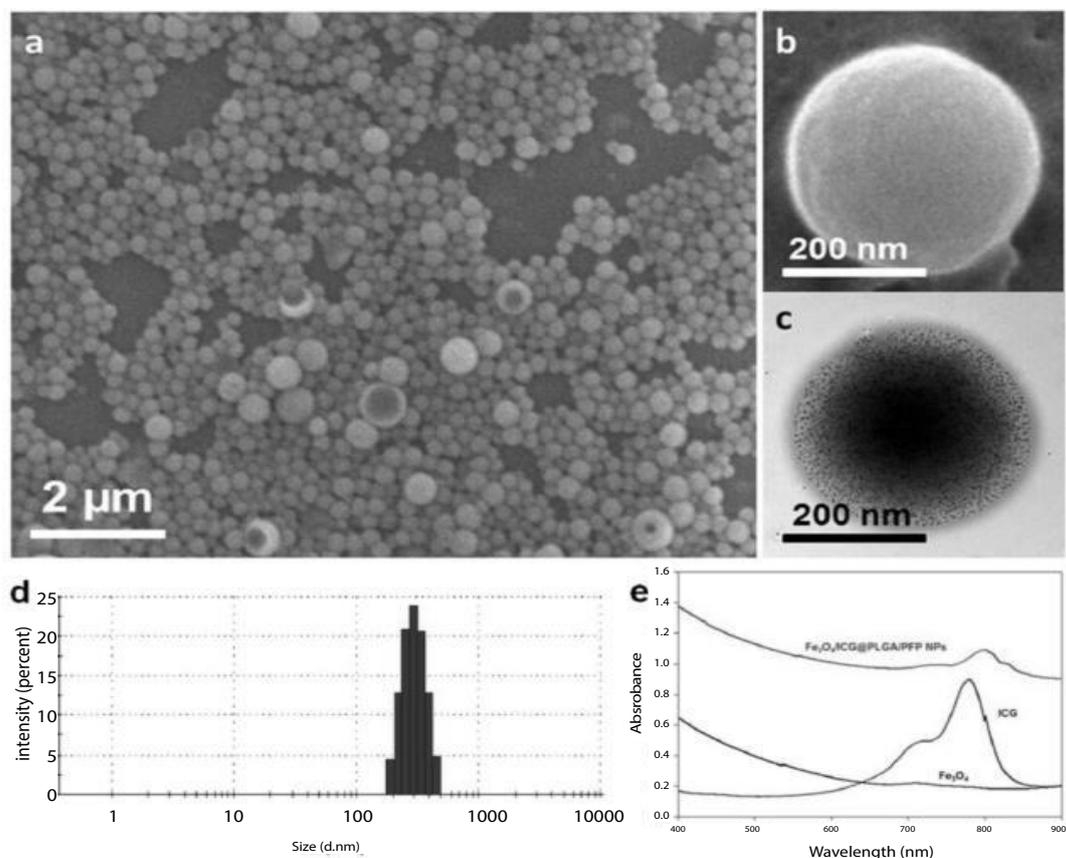


Figure 5: TEM image of Croc/LP-hybrid-NP.

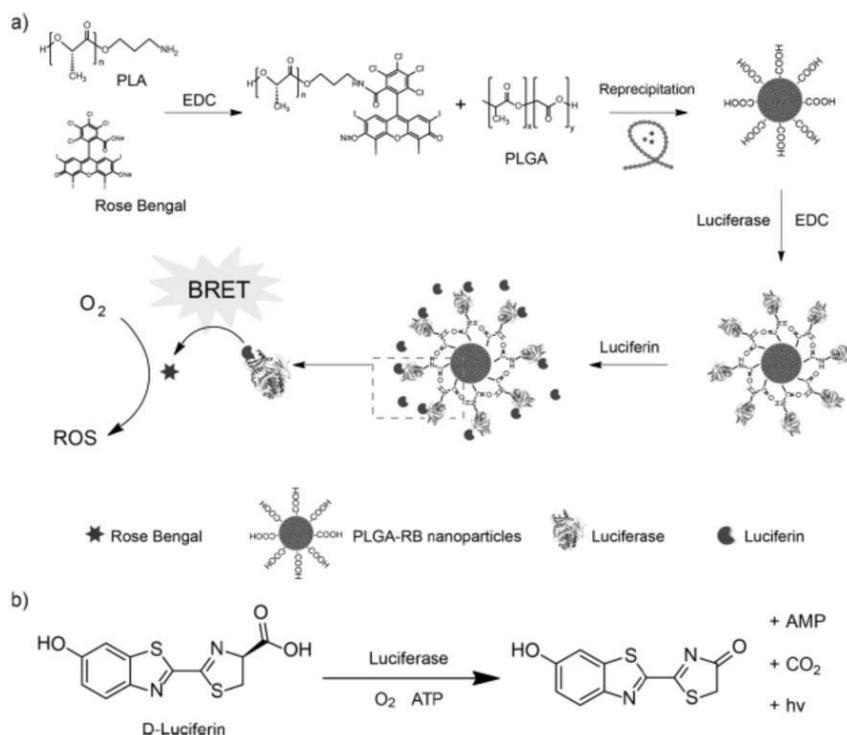
Near-infrared (NIR) laser-induced photothermal therapy (PTT) uses a photothermal agent to convert optical energy into thermal energy and has great potential as an effective local, minimally invasive treatment modality for killing cancer cells. To improve the efficacy of PTT, Chengcheng Niu et al. [35] developed poly(lactide co-glycolide) (PLGA) nanoparticles (NPs) encapsulating superparamagnetic iron oxide ( $\text{Fe}_3\text{O}_4$ ), indocyanine green (ICG), and perfluoropentane (PFP) as synergistic agents for NIR laser-induced PTT. They fabricated a novel type of phase-shifting fluorescent magnetic NPs,  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs, that effectively produce heat in response to NIR laser irradiation for an enhanced thermal ablation effect and a phase-shift thermoelastic expansion effect, and thus, can be used as a photothermal agent. Figure 6

After *in vitro* treatment of MCF-7 breast cancer cells with  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs and NIR laser irradiation, histology and electron microscopy confirmed severe damage to the cells and the formation of many microbubbles with iron particles at the edge or outside of the microbubbles. *In vivo* experiments in mice with MCF-7 tumors demonstrated that  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs could achieve tumor ablation upon NIR laser irradiation with minimal toxicity to non-irradiated tissues. Together, their results indicate that  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs can be used as effective nanotheranostic agents for tumor ablation. Hrebesh M. Subhash, et al. [36] have been described a functional

imaging paradigm that uses photothermal optical coherence tomography (PT-OCT) to detect indocyanine green (ICG)-encapsulated biocompatible poly(lactic-co-glycolic) acid (PLGA) nanoparticles embedded in highly scattering tissue phantoms with high resolution and sensitivity. The ICG-loaded PLGA nanoparticles were fabricated using a modified emulsification solvent diffusion method. With a 20 kHz axial scan rate, PT-OCT based on spectral-domain interferometric configuration at 1310 nm was used to detect phase changes induced by a 808 nm photothermal excitation of ICG-encapsulated PLGA nanoparticles. An algorithm based on Fourier transform analysis of differential phase of the spectral interferogram was developed for detecting the depth resolved localized photothermal signal. Excellent contrast difference was observed with PT-OCT between phantoms containing different concentrations of ICG-encapsulated PLGA nanoparticles. This technique has the potential to provide simultaneous structural and molecular-targeted imaging with excellent signal-to-noise for various clinical applications. Photoacoustic imaging (PAI) is an emerging biomedical imaging technique that is now coming to the clinic. It has a penetration depth of a few centimeters and generates useful endogenous contrast, particularly from melanin and oxy-/deoxyhemoglobin. Indocyanine green (ICG) is a Food and Drug Administration-approved contrast agents for human applications, which can be also used in PAI. It is a small molecule dye with limited applications due to its fast clearance, rapid protein binding, and bleaching effect.



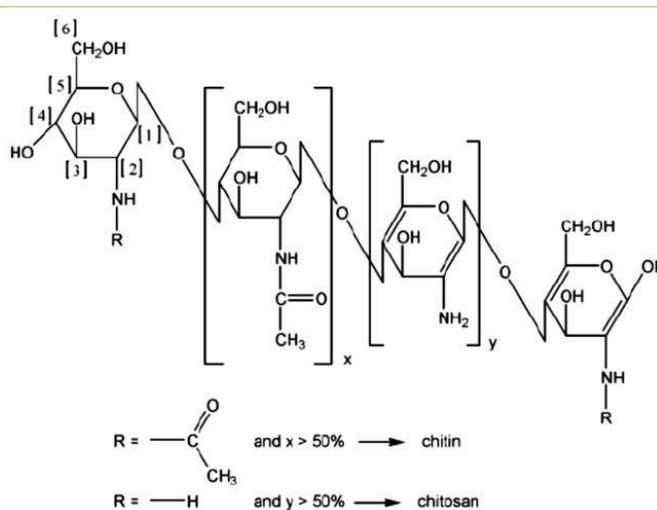
**Figure 6:** (a,b) SEM images of  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs (a) 10000 $\times$ ; scale bar, 2  $\mu\text{m}$ ; (b) 70000 $\times$ ; scale bar, 200 nm); (c) TEM image of  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs with a large amount of black  $\text{Fe}_3\text{O}_4$  NPs embedded in the spherical shell (70000 $\times$ ; scale bar, 200 nm); (d) size distribution of  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs with the average diameter of  $289.6 \pm 67.4$  nm; (e) UV-Vis-NIR absorption spectra of  $\text{Fe}_3\text{O}_4/\text{ICG}@PLGA/\text{PFP}$  NPs, free ICG and  $\text{Fe}_3\text{O}_4$  NPs.



**Figure 7:** (a) Schematic illustration of the preparation of PLGA-RB nanoparticles. (b) Firefly luciferase catalyzing the oxidation reaction of the native substrate D-luciferin for photon emission

Edyta Swider et al. [37] showed that they can label primary human dendritic cells with the nanoparticles and image them *in vitro* and *in vivo*, in a multimodal manner. This work demonstrates the potential of combining PAI and <sup>19</sup>F MRI for cell imaging and lymph node detection using nanoparticles that are currently produced at GMP-grade for clinical use. Photodynamic therapy (PDT) is emerging as a noninvasive modality for a variety of cancer treatments [38,39]. In typical PDT, a photosensitizer is activated by external light to produce reactive oxygen species (ROS), which can damage cancer cells [40-42]. For deep-seated tumors, the efficiency of PDT is decreased due to the limited penetration depth of external light in biological tissue. This problem has hindered further development of PDT in widespread clinical applications. In addition, skin photosensitivity and photothermal injury are also common concerns from the patients receiving PDT treatment [43-45]. For the light penetration issue to be overcome, considerable research has focused on the development of photosensitizers in the near infrared (NIR) range [46]. Up-conversion nanoparticles combined with photosensitizers have been explored because they can absorb NIR photons and emit visible light to activate the photosensitizer for PDT [47,48]. Researchers have explored new approaches, for example, using Cerenkov radiation to activate a photosensitizer for effective PDT [49,50]. Alternatively, X-ray is currently widely used in clinical treatment, and X-ray-activated photodynamic therapy is another way to overcome the light penetration limitation in deep-seated tumors [51-53]. The use of an internal light source is an intriguing solution for the light-penetration problem. For example, implanting fiberoptic light sources and emitting diodes could be a viable approach, but it still requires invasive operation [54-56]. Alternatively, bioluminescence offers an attractive strategy

because of the minimally invasive delivery of enzymes and substrates. Bioluminescence is a natural phenomenon that occurs in various organisms, such as marine organisms and some insects, in which visible light is produced via chemical reactions *in vivo* [57,58]. Bioluminescence has been widely used in biological detection and optical imaging [59-61]. Recently, researchers have recognized the therapeutic potential of bioluminescence. Bioluminescence is endogenous fluorescence that can be used to activate the photosensitizer inside the tumors. This process is not affected by the light penetration depth of tissue. Wang et al. reported that luminol has anticancer and antifungal activities through bioluminescence resonance energy transfer (BRET) [62]. Moreover, quantum dots have also been combined with the BRET system for *in vivo* imaging and PDT in recent reports



**Figure 8:** Chemical structure of Chitin and Chitosan.

[63-65]. Despite these progresses, the BRET-activated ROS generation and photodynamic effect is largely unexplored. They show effective photodynamic therapy mediated by the firefly luciferase bioluminescence system. Biodegradable poly(lactic-co-glycolic acid) (PLGA) nanoparticles were doped with the photosensitizer Rose Bengal (RB). The PLGA-RB nanoparticles were then conjugated with luciferase, which produced a fluorescent signal in the presence of the luciferin substrate. The endogenous bioluminescence served as a light source to activate the photosensitizer for ROS generation. They evaluated cell toxicity, cellular fluorescence imaging, and therapeutic effect in cellular assays, which indicated that cancer cells were destroyed effectively. The antitumor effect of BRET-PDT was further investigated in tumor-bearing ICR mice, which showed that tumor growth was significantly inhibited. This study provides a promising approach for phototherapy of deep-seated tumors in the absence of external excitation. Figure 7

Nanoparticles present great potential in drug delivery applications, yet there are some issues regarding their stability. In this study was conducted to define the conditions to stabilize polysaccharide (chitosan/dextran sulfate, CS/DS) nanoparticles by a process of freeze-drying, assessing the cryoprotectant capacity of two sugars (sucrose and glucose). Additionally, it was also intended to find if the solubilisation of chitosan in different acids affected nanoparticle preparation and characteristics. Chitosan is obtained by the *N*-deacetylation of chitin, a polymer that can be extracted from various sources (crustacean's shells, exoskeletons of certain invertebrates – ladybugs – and the cell walls of fungi, for example). It is considered the second most abundant polymer on Earth [66], after cellulose [67]. However, chitin is not very versatile due to its structure and poor solubility in many solvents. Chitin plays a role in the protection of certain animals in nature, being organized in semi-crystalline microfibrils to provide the said protection [68]. The deacetylated form of chitin is chitosan, which structure can be seen on Figure 8.

Chitosan (MW ~30-190 kDa) is the polymer obtained when deacetylation surpasses 50% and is comprised of  $\beta$  (1-4)-links of D-glucosamine and *N*-acetyl-D-glucosamine monomers that are distributed randomly throughout the chain. This is the only natural polymer exhibiting a cationic character [69]. Due to the fact that the amine groups are present throughout the structure of chitosan, the more deacetylated the polymer is, the more susceptible it is to protonation because nitrogen has an unused pair of electrons that can easily interact with electrophilic groups. When chitosan undergoes protonation, which occurs at low pH levels, it acquires a positive charge, thus providing the possibility to interact with negatively charged groups. This ability has been widely explored in drug delivery, with the preparation of nanoparticles by electrostatic interaction, as reported using carrageenan [70], tripolyphosphate (TPP) or dextran sulfate [71] as counterions. Chitosan is not soluble in water. Instead, and because the amine groups have a logarithmic acidity constant (pKa) of ~6.5, chitosan dissolves easily in acidic media. The most usual solvent for its dissolution is 1% (v/v) acetic acid. Due to the fact that the polymer is easily protonated, it can also be dissolved in

formic and lactic acids and in hydrochloric acid solutions [72]. In a pH of 7 or higher, as the pH is higher than the pKa of amine groups, chitosan becomes insoluble, rendering its biological applications are scarce or non-existent. As mentioned earlier, chitosan presents adequate characteristics regarding biological applications [73], including biocompatibility, biodegradability and low toxicity. Its biodegradability is due to a metabolism by lysozyme [74]. Moreover, due to its protonation in acidic media, it presents a favourable interaction with mucus [75], which is particularly relevant in the intestine, where there is a high amount of mucus. This is actually one of the reasons justifying its application in specific disease conditions, such as colonic inflammation. Nevertheless, the description of chitosan applications in literature is endless, including pulmonary and oral insulin delivery as well as the delivery of antibiotics and small genetic fragments [76-78], just to mention some examples. Finally, chitosan can also be found in a variety of dietary supplements (Lipoforte® and EasySlim® Blocker, just to name a few) for weight loss. However, the European Food Safety Authority (EFSA) considers that there is not enough evidence that supports this indication. There are other studies that report its beneficial use on lowering LDL-cholesterol [79,80], a claim that EFSA found to have enough evidence that supports it. However, some problems have also been described for supplements, reporting chitosan interference with certain treatments, for example, anticoagulation therapy with warfarin [81] and anti-epilepsy therapy with valproate [82]. Dextran sulfate was the other polymer used in this experimental work. It is also a natural polymer, obtained from bacteria of the *Leuconostoc* genus. It is comprised of a branched anhydroglucose backbone with attached sulphur groups, the latter believed to be 17% of the total mass of the polymer. This means that, 2.3 sulphur groups are present for each glucosyl residue, thus the polymer being expected to be negatively charged. Its chemical structure can be seen in Figure 2. Dextran sulfate (MW ~5-500 kDa [83-85] is also reported as biocompatible and biodegradable [86], thus theoretically complying with mandatory requisites for biomedical applications. Studies regarding the use of dextran sulfate as a component of DDS are scarce, even though it was widely studied in the eighties for its anti-HIV activity, against HIV-1. Its mechanism of action, in this context, involves inhibiting the reverse transcriptase, a viral enzyme responsible for turning the viral RNA in viral DNA that will, ultimately, be part of the host cell DNA resulting in the production of viral particles [87]. Additionally, dextran sulfate has also been shown to activate macrophages by establishing an interaction with the scavenger receptor present in these cells [88]. Moreover, regarding the immune system, it is believed that dextran sulfate induces activation of B lymphocytes even though that mechanism is mediated by macrophages [89]. Another aspect that is widely mentioned in studies is an anticoagulant capacity similar to heparin, even though it wasn't studied further [90]. Finally, dextran sulfate is used, nowadays, to induce colitis in mouse models [91,92]. Approved formulations containing this polymer include two EMA-approved orphan medicines: i) a medicine for the prevention of graft rejection and its further damage on pancreatic

transplantation by inhibiting the activation of complement system and clotting [93] and ii) a medicine to promote mobilisation of progenitor cells, prior to stem cell transplantation, through a cytokine released by bone marrow that can guide these important cells through blood stream [94]. Dextran sulfate is a very interesting polymer with many characteristics that can be used in drug delivery. However, there is still much work to do and, more importantly, study possible associations with other polymers to produce nanocarriers with a potential use in therapeutics [95-97]. Chitosan extracted from *Metacarcinus magister* shells using demineralization with 1.5M and 2M HCl showed the characteristic functional groups of CS. The CS-CMC nanoparticle were chelated with five different concentrations of TPP and  $\text{BaCl}_2$  (0.1%, 0.2%, 0.3%, 0.4% and 0.5%) and embodied with CUR. The SEM analysis indicated the size of the nanoparticles to below 500 nm which was also confirmed by AFM. The encapsulation efficiency was absolutely time dependent. The drug release was found to be best with higher concentration of chelating agents used and the drug release was prolonging to 80th minute for all the samples. In vitro drug release showed antibacterial activity when ethanol and acetic acid was used as solvent system [98]. A new granule hydrogel composites of natural biocompatible polymeric chitosan and gelatin (C-G) have prepared. The gelatin side groups (hydroxyl and carboxyl) can attach with amine groups of chitosan, which improve the interlink crosslinking in the final nanocomposites. The diameter of the granules is around 3mm to 6 mm. Furthermore, in order to reinforce the mechanical properties of granules, for the first time, octa(ammonium chloride) substituted Polyhedral Oligomeric Silsesquioxane C-G/POSS-( $\text{NH}_3+\text{Cl}$ )<sub>8</sub> has been used as a nanofiller in the structure of this nanocomposite. The effect of C-G/POSS-( $\text{NH}_3+\text{Cl}$ )<sub>8</sub> on mechanical stability and swelling behavior of hydrogels has been investigated in three pH of 1.2, 7.4 and 9.5, respectively. Results demonstrate that the C-G/POSS-( $\text{NH}_3+\text{Cl}$ )<sub>8</sub> granules in acidic pH has the maximum swelling percent. The encapsulation efficiency (EE) values for C-G granules in three pH of 1.2, 7.4 and 9.5 are obtained 65%, 58% and 53% and for C-G/POSS-( $\text{NH}_3+\text{Cl}$ )<sub>8</sub> sample are 59%, 52% and 48%, respectively, which conform with the swelling behavior of these hydrogels. According to the obtained results from Metronidazole (MTZ) release, the as-synthesized hydrogels may have intrinsic ability in the controlled release of drugs. The as-prepared hydrogels are characterized via FT-IR, SEM, TGA and XRD analysis [99]. Chitosan nanoparticles (CS NPs) exhibit good physicochemical properties as drug delivery systems. Haliza Katas and et al. [100] have determined the modulation of preparative parameters on the physical characteristics and colloidal stability of CS NPs. CS NPs were fabricated by ionic interaction with dextran sulphate (DS) prior to determination of their storage stability. The smallest CS NPs of  $353 \pm 23 \text{ nm}$  with a surface charge of  $+56.2 \pm 1.5 \text{ mV}$  were produced when CS and DS were mixed at pH 4 and with a DS : CS mass ratio of 0.5 : 1. An entrapment efficiency of 98% was achieved when BSA/siRNA was loaded into the nanoparticles. The results also showed that particle size and surface charge of CS NPs were slightly changed up to 2 weeks when stored at 4°C. Greater particle size and surface charge were obtained

with increasing the concentration of DS. In conclusion, NPs were sufficiently stable when kept at 4°C and able to carry and protect protein. Ji Woon Suh and et al. [101] have formulated buccal mucoadhesive nanoparticles (NPs) using the natural mucoadhesive polymers. The natural mucoadhesive polymers chitosan (CS) and dextran sulfate sodium salt (DS) were used to prepare mucoadhesive NPs using the ionic gelation method. As the molecular weight of DS decreased, the amount of mucin and the number of buccal cells adsorbed on DS increased. The CS/DS NPs ranged from 100 to 200 nm in diameter. The adhesive interactions of CS/DS NPs with mucin were not significantly different from those of CS/sodium triphosphate pentabasic (TPP) NPs; however, CS/DS NPs exhibited 5 times greater mucoadhesive activity to buccal cells compared to control CS/TPP NPs in ex vivo adhesion tests. These results indicate that the buccal mucoadhesive properties of NPs can be improved using natural mucoadhesive polymers. A novel nanoparticle delivery system has been developed by employing the oppositely charged polymers chitosan (CS) and dextran sulfate (DS), and a simple coacervation process. Under the conditions investigated, the weight ratio of the two polymers is identified as a determining factor controlling particle size, surface charge, entrapment efficiency and release characteristics of the nanoparticles produced. Particles of 223 nm mean diameter were produced under optimal conditions with a zeta potential of approximately -32.6 mV. A maximum of 75% anti-angiogenesis peptide entrapment efficiency was achieved with a CS:DS weight ratio of 0.59:1. The same nanoparticle formulation also showed slow and sustained peptide release over a period of 6 days. In contrast, the formulation containing a lower ratio of CS:DS (0.5:1) was found to have reduced entrapment efficiency and more rapid peptide release characteristics. The results suggest that physicochemical and release characteristics of the CS-DS nanoparticles can be modulated by changing ratios of two ionic polymers. The novel CS-DS nanoparticles prepared by the coacervation process have potential as a carrier for small peptides [102]. Fwu-Long Mi [40] have prepared chitosan-based nanoparticles containing indocyanine green (ICG) for hyperthermia therapy and imaging. ICG was incorporated into chitosan and poly- $\gamma$ -glutamic acid self assembled nanoparticles using the polyelectrolyte complex method. The ICG encapsulation efficiency of the nanoparticles were higher than 95% and the ICG-loaded nanoparticles showed effective temperature elevation upon NIR laser irradiation (808 nm, 1.5 W/cm<sup>2</sup>). The cytotoxicity results from NIR laser-induced hyperthermia effect showed an enhanced toxicity in overcoming doxorubicin (Dox) resistance of MCF-7/ADR cells. Pei-Ru Wei et al. [103] have designed a study for photodynamic therapy (PDT) using chitosan coated Mg-Al layered double hydroxide (LDH) nanoparticles as the delivery system. A Food and Drug Administration (FDA) approved near-infrared (NIR) fluorescent dye, indocyanine green (ICG) with photoactive properties was intercalated into amine modified LDH interlayers by ion-exchange. The efficient positively charged polymer (chitosan (CS)) coating was achieved by the cross linkage using surface amine groups modified on the LDH nanoparticle surface with glutaraldehyde as a spacer. The unique hybridization of

organic-inorganic nanocomposites rendered more effective and successful photodynamic therapy due to the photosensitizer stabilization in the interlayer of LDH, which prevents the leaching and metabolization of the photosensitizer in the physiological conditions. The results indicated that the polymer coating and the number of polymer coats have a significant impact on the photo-toxicity of the nano-composites [104-106]. The double layer chitosan coated LDH-NH<sub>2</sub>-ICG nanoparticles exhibited enhanced photo therapeutic effect compared with uncoated LDH-NH<sub>2</sub>-ICG and single layer chitosan-coated LDH-NH<sub>2</sub>-ICG due to the enhanced protection to photosensitizers against photo and thermal degradations. This new class of organic-inorganic hybrid nanocomposites can potentially serve as a platform for future non-invasive cancer diagnosis and therapy. Nanoparticles (NPs) are synthetic materials with dimensions from one to hundreds of nanometers, and remarkable applications in biomedicine due to the unique way in which they interact with matter [107,108]. There are currently more than 35 US FDA-approved NPs often incorporating polyethylene glycol (PEG), with a larger number in preclinical studies for both imaging and therapy (Figure 9A) [109–115]. NPs have large payloads, stability, avidity, signal enhancement and the capacity for multiple, simultaneous applications owing to their unique size and high surface area:volume ratio [116]. While they are bigger than molecules and many proteins, yet smaller than cells, they behave differently to other therapies and imaging agents, affecting their *in vivo* applications. For example, in cancer tissue, NPs not only extravasate from the leaky tumor vasculature to a higher degree than healthy tissue, but also remain in the area by the enhanced permeability and retention (EPR) effect [117]. NPs lodged in the tumor can then perform signaling and/or therapy [118]. Despite these advantages, some fundamental challenges hamper NP deployment to the clinic. These include uptake by the reticuloendothelial system (RES), in which NPs are rapidly shuttled out of circulation to the liver, spleen or bone

marrow, and nonspecific binding of NPs to nontargeted or nondiseased areas. Concerns about NP toxicity often arise because of this RES accumulation. Aggregation can lead to NP entrapment in the liver, lungs or elsewhere due to capillary occlusion [119]. The addition of PEG to the NP surface (PEGylation) can reduce many of these challenges (Figure 1B). PEG is a coiled polymer of repeating ethylene ether units with dynamic conformations (Figure 1C). In both drug-delivery and imaging applications, the addition of PEG to NPs reduces RES uptake and increases circulation time versus uncoated counterparts [120]. Aggregation decreases owing to passivated surfaces, and association with nontargeted serum and tissue proteins is diminished, resulting in so-called ‘stealth’ behavior. The PEG chains reduce the charge-based contact typical of proteins and small-molecule interactions. Solubility in buffer and serum increases due to the hydrophilic ethylene glycol repeats and the EPR effect is modulated due to NP size changes via addition of a PEG coat [121,122]. Due to these attributes, PEGylated NPs generally accumulate in the liver a half to a third of the amount of non-PEGylated NPs and demonstrate higher tumor accumulation versus background [123]. PEG is inexpensive, versatile and FDA approved for many applications [124]. All NPs contain at least two fundamental spatial components: the core and the corona that interact with the environment or solvent. While core/shell, core/multishell systems add further complexity, for example [125], all still possess an area in which NP interfaces with the solvent (Figure 9B). PEG chains modify this interface layer and increase circulation time. Circulation half-time ( $t_{1/2}$ ) describes blood pool residence and is the period over which the concentration of circulating NPs remains above 50% of the injected dose, analogous to a drug’s half-life [126]. NP efficacy requires sufficient  $t_{1/2}$  to not only reach the target, but also remain in the affected area (at concentrations sufficiently above background tissue) long enough for image capture or drug delivery. The RES system prevents site-specific accumulation because it removes the NPs from

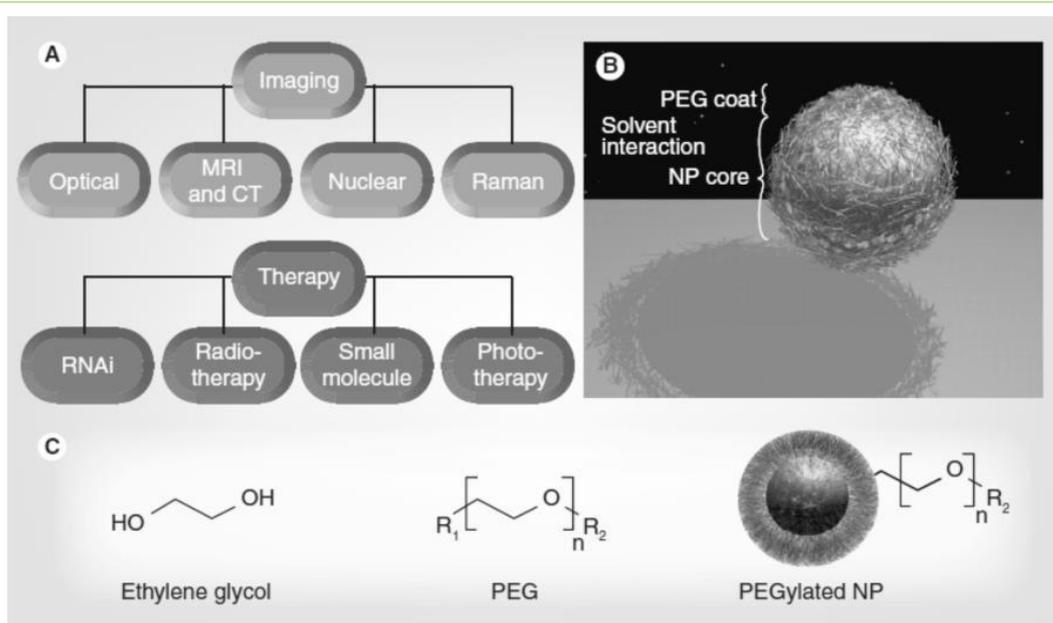
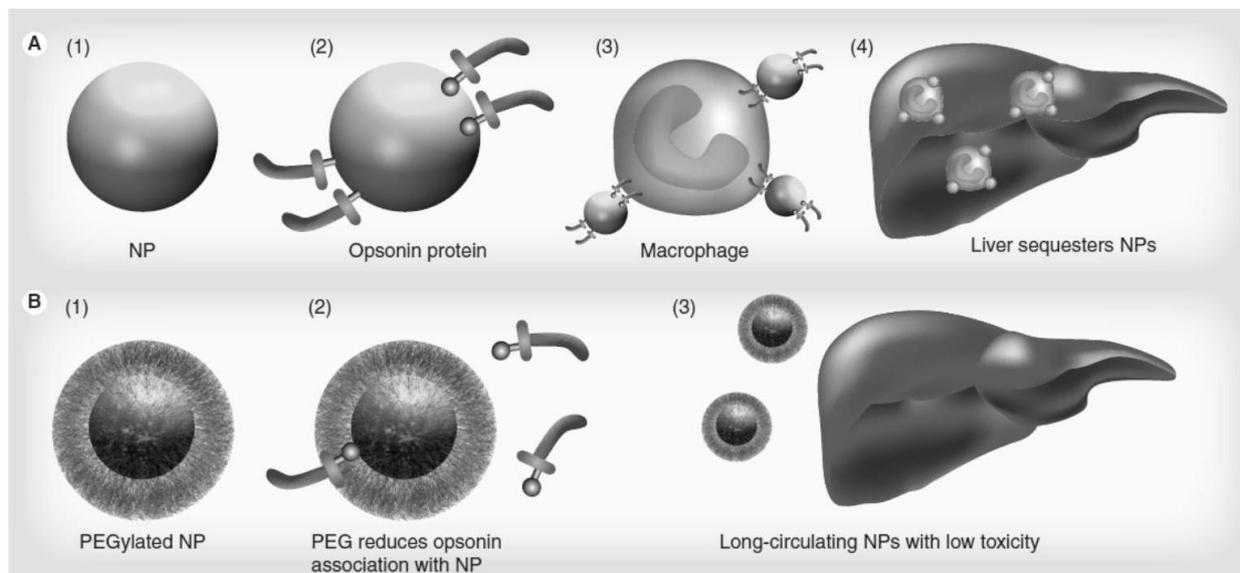


Figure 9: Nanoparticle applications of Polyethylene glycol.



**Figure 10:** Polyethylene glycol prevents uptake by the reticuloendothelial system.

circulation, acting as a competitor to the intended target site [127]. In addition, the NPs must clear from the nontargeted area to produce imaging contrast or dosing efficiency. The ideal  $t_{1/2}$  is dependent on application. In imaging, 2–6 h is optimal for injection, accumulation at targeted site, clearance from nontargeted areas and data collection. The ideal circulation time for therapeutic NPs is longer (days) to allow repeated exposure to affected area. Unfortunately, this can also expose healthy organ systems to the drug and is the motivation for targeted NPs, as such systems preferentially accumulate in the diseased area. Approaches to measuring  $t_{1/2}$  vary with NP type. When labeled with radionuclides, g counting of either specific organ systems or blood aliquots determines NP circulation time. One limitation is dissociation of radionuclide from NPs; however, radioactivity measurements may always be carried out noninvasively [128]. Figure 9

Measurement of  $t_{1/2}$  via fluorescence, Raman, inductively coupled plasma or chromatography/mass spectrometry is very specific to the NP, but requires sequential sampling of the blood pool. The RES is an immune system component, utilizing circulating macrophages and monocytes, liver Kupffer cells and spleen and other lymphatic vessels to remove foreign material, such as bacteria and viruses, from the body [127]. Figure 10 illustrates how opsonin proteins associate with foreign bodies and coat its surface [128]. As bacteria and viruses have the same negative surface charge as phagocytic cells, opsonins are critical to reducing the charge repulsion between the two systems. Next, phagocytic cells engulf the material and transport it to the liver or spleen for degradation and excretion (Figure 10 A3–A4). Additional phagocytic macrophages are permanently located in the liver. Known as Kupffer cells, these cells serve as a major filter for many types of NPs and are a major interference with long  $t_{1/2}$  [129]. The PEG polymer on a NP surface increases  $t_{1/2}$  by reducing this opsonization process (Figure 10B2), thus preventing recognition by monocytes and macrophages, allowing the NPs to remain in the blood pool. Hydrophobic particles are also more vulnerable to the RES

and hydrophilic PEG reduces these complications. In addition to NP–RES interactions, poor  $t_{1/2}$  can also result from NP–NP interactions (i.e., aggregation). NPs aggregate primarily because the attraction between particles is stronger than the attraction for solvent [130]. NPs with a high surface energy have a greater tendency to aggregate as described by the Derjaguin–Landau–Verwey–Overbeek (DLVO) theory [131,132]. For spherical NPs, the interaction potential is related to the electrostatic repulsive potential and the van der Waals attraction potential [133]. PEG decreases the surface energy of NPs and minimizes van der Waals attraction [134–135]. Aggregation can be induced by solvents of high (>100 mM) ionic strength (shielding of solvent from NP), highly concentrated solutions of NPs (less distance between the NPs), time from synthesis, or NP preparations with a very neutral ( $\sim\pm 5$  mV) zeta potential [136]. PEG decreases the amount of attraction between NPs by increasing the steric distance between them and increasing hydrophilicity via ether repeats forming hydrogen bonds with solvent. Other benefits to PEGylation include modifying the size of the particle. The reduced renal filtration of particles larger than 10 nm increases  $t_{1/2}$ ; however, at too large a size (>100 nm), liver uptake increases and EPR extravasation may decrease [137]. PEG modifies the NP flexibility and the NP can become ‘softer’ after PEGylation than the underlying material, influencing extravasation.

Prior to NP applications, PEG was used as a nontoxic, water-soluble dispersant/stabilizer. Also known as Carboxwax®, it is present in health and beauty aids, including laxatives, toothpastes and eye drops, and is an excipient in tablet formulations [138]. PEG stabilizes organ and blood donations. Early work with PEGylated NPs stemmed mostly from drug delivery [139]. One of the first reports on PEGylation was described by Davis and Abuchowski [140,141], where they covalently attached methoxy-PEGs (mPEGs) of 1900 and 5000 Da to bovine serum albumin and to liver catalase. Later, acrylic microspheres functionalized with PEG-modified human serum albumin increased  $t_{1/2}$  *in vivo* [142]. Li and colleagues found that

75-nm latex particles remained in rat circulation 40-times longer (half-life 20 min vs 13 h) when coated than uncoated with PEG larger than 5000 kDa [143]. Klibanov and Huang found that incorporation of dioleoyl *N*- (monomethoxy polyethyleneglycol succinyl) phosphatidylethanolamine (PEG-PE) into phosphatidylcholine:cholesterol liposomes (1:1) increased  $t_{1/2}$  from 30 min to 5 h without increasing leakage of the liposome interior. In the mid-1990s, Doxil® (liposomal delivery vehicle for doxorubicin) and oncospar (PEG-l-asparaginase) became the first FDA-approved NP therapeutics [143]. Doxil increases doxorubicin bioavailability nearly 90-fold at 1 week from injection of PEGylated liposomes versus free drug [144]. The use of PEG on the doxorubicin carrier yields a drug half-life of 72 h with circulation half-life of 36 h [145,146]. Later, Abraxane® was introduced as an albumin-functionalized NP for delivery of taxane without cremphor to enhance drug efficiency. Thadakapally, *et al.* developed a novel serum stable long circulating polymeric nanoparticles for curcumin with a modification to the well known and novel nanoparticle albumin bound technology. polyethylene glycol-albumin-curcumin nanoparticles were prepared using serum albumin and poly ethylene glycol using desolvation technique. Nanoparticles were characterized for encapsulation efficiency, particle size and surface morphology. Drug excipient compatibility was determined using fourier transform infrared spectroscopy. Physical state of the drug in the formulations was known by differential scanning calorimetry. *In vitro* release and solubility of the drug from nanoparticles were determined. *In vivo* Drug release, tissue uptake and kupffer cell uptake was determined with optimized nano formulation in rats after intravenous administration. Cell viability assay was determined using breast cancer cell line MD-MB-231. Entrapment efficiency for prepared nanoparticle was above 95%. The polyethylene glycol-albumin-curcumin nanoparticles exhibited an interesting release profile with small initial burst followed by slower and controlled release. Solubility of the drug from the formulation was increased. A sustained release of drug from nanoparticles was observed for 35 days in both *in vitro* and *in vivo* studies with the optimized formulation. Polyethylene glycol-albumin-curcumin nanoparticles showed lesser liver and kupffer cell uptake as compared to that of curcumin-albumin nanoparticles suggesting the bestowment of stealthness to nanoparticles with pegylation. Also, the antiproliferative activity of polyethylene glycol-albumin-curcumin nanoparticle formulation was more as compared to native curcumin. Polyethylene glycol-albumin-curcumin nanoparticles thus developed can be conveniently used in breast cancer with improved efficacy compared to conventional therapies and as an alternate to nanoparticle albumin bound technology which is used in producing Abraxane, albumin based breast cancer targeting nanoparticles of paclitaxel [147]. Photothermal therapy (PTT) and photodynamic therapy (PDT) are emerging physical tumor treatments utilizing near infrared (NIR) light-absorbing agents which lead to thermal ablation of cancer cells or generate highly reactive oxygen species (ROS) via photosensitizer to ablate tumors [148,149]. PTT and PDT possess several advantages, such as minimal invasion, high

therapeutic efficacy, limited side-effects, selective localized treatment and reproducible properties [150,151], and hence have received much attention in recent years [152,153]. Until now, a variety of materials has been explored as PTT or PDT agents due to their high absorption in the tissue-transparent NIR wavelength range, including organic fluorescent dyes [154], gold nanorods [155], CuS nanoparticles (NPs) [156], polymer NPs [157], carbon nanomaterials [158], etc. [159,160]. However, fluorescent dyes may be removed rapidly from the systemic circulation and lack specificity to a tumor, and inorganic photothermal agents have potential long-term toxicity due to the difficulty of degrading in the body [161]. Therefore, exploiting biocompatible and targeted therapeutic nanoagents with enhanced photothermal conversion capability and ROS generation ability to amplify PTT and PDT treatments remains challenging. Indocyanine green (ICG) is a clinical infrared imaging agent approved by the U.S. Food and Drug Administration (FDA), and has been applied in optical imaging of human vasculature, tissue and cells due to its biocompatibility and unique optical properties [162]. Due to strong absorption at 780 nm, ICG can effectively convert absorbed NIR optical energy into heat for PTT [163] and produce ROS for PDT [164] under NIR laser irradiation [165]. Nevertheless, the application of ICG in tumor phototherapy is limited by its tendency to aggregate, rapid degradation in aqueous solution [64], poor photo-stability and non-specific binding to proteins [166]. To overcome those limitations, various nanoparticle delivery systems have been developed to encapsulate ICG [167]. Lvet al. used a mesoporous silica (mSiO<sub>2</sub>) matrix to load ICG molecules, and demonstrated that loaded ICG displayed a more enhanced photothermal effect than pure ICG [168]. ICG-loaded mesoporous silica NPs also could not only limit the degradation of ICG, but reach and stay at a tumor for a long period of time due to an enhanced permeability and retention (EPR) effect [169]. Hence, loading of ICG within targeting nanocarriers with high efficiency is shown to be an effective way to promote the application of ICG in PTT and PDT treatment.

## Conclusion

Multifunctional biodegradable and biocompatible polymeric nanoparticles loaded with indocyanine green (ICG) as an optical-imaging contrast agent for cancer imaging and as a photothermal therapy agent for cancer treatment. To improve the efficacy of PTT, polymeric nanoparticles (NPs) encapsulating superparamagnetic iron oxide (Fe<sub>3</sub>O<sub>4</sub>), indocyanine green (ICG), and perfluoropentane (PFP) as synergistic agents for NIR laser-induced PTT.

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